

Chemistry 206 (Organic Chemistry) Syllabus and Schedule

Basic Course Information

Semester:	Spring 2023	Instructor Name:	Dr. Alto Benedicto
Course Title & #:	Chemistry 206	Email:	alto.benedicto@imperial.edu
CRN #:	20539	Units:	5
Classroom:	online	Office #:	2779 online
Class Dates:	Feb 13 to Jun 9, 2023	Office Hours:	MTWTh 6:30 am – 7:30 am Zoom
Class Days:	Tutor available MTWThF (see page 5)	Office Phone #:	(760) 355-5751
Class Times:	online	Emergency Contact:	Department Secretary (760) 355-6155

Course Description

This course is a study of various reactions and properties aldehydes, ketones, carboxylic acids, aromatic compounds, amines, conjugated dienes, lipids, carbohydrates, and organic polymers. A survey of various biochemical topics such as metabolism, protein structure, and DNA is also included. This course is a continuation of CHEM 204 and is intended for students majoring in chemistry, biology, and pre-medical sciences. (CSU, UC)

Student Learning Outcomes

Upon course completion, the successful student will have acquired new skills, knowledge, and or attitudes as demonstrated by being able to:

1. Solve chemical problems using modern atomic theory (ISLO 2, ISLO 4)
2. Perform chemical experiments in a scientific manner, using proper techniques, analysis, and safety equipment. (ISLO 2, ISLO3, ISLO4)

Course Objectives

Upon satisfactory completion of the course, students will be able to:

1. demonstrate knowledge of the structure and reactions aldehydes and ketones.
2. demonstrate knowledge of the structure and reactions carboxylic acids and their derivatives.
3. demonstrate knowledge of enolate anions and enamines.
4. demonstrate knowledge of the structure and reactions of aromatic compounds.
5. demonstrate knowledge of the structure and reactions of amines.
6. demonstrate knowledge of the structure and reactions of conjugated dienes.
7. demonstrate knowledge of organic polymers.
8. demonstrate knowledge of the structure and reactions carbohydrates.
9. demonstrate knowledge of lipids.
10. demonstrate knowledge of the chemistry of metabolism.
11. demonstrate knowledge of the structure and reactions of amino acids and proteins.
12. demonstrate knowledge of nucleic acids and DNA.

Textbooks & Other Resources or Links

REQUIRED MATERIALS:

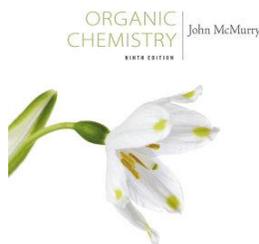


Figure 1: Organic Chemistry book by John McMurry, 9th edition

1. *Organic Chemistry*, by John McMurry. Cengage Learning, 9th Ed, ISBN: 9781337158459 (see #2 on how to purchase)
NOTE: There is SHORTENED ONLINE version at [https://chem.libretexts.org/Bookshelves/Organic_Chemistry/Map%3A_Organic_Chemistry_\(McMurry\)](https://chem.libretexts.org/Bookshelves/Organic_Chemistry/Map%3A_Organic_Chemistry_(McMurry))
2. Enroll into OWL2 via <https://www.cengage.com/dashboard/#/course-confirmation/E-KY655K698WRZR/initial-course-confirmation> / [Student Registration Instructions](#) This purchase includes the following: *OWL2 online HW*, *LabSkills prelab*, digital copy of *Organic Chemistry*, by John McMurry of the book. You may also purchase all these through the IVC bookstore. Course Key is **E-KY655K698WRZR**
3. Select Experiments in *A Small Scale Approach to Organic Laboratory Techniques*, by Donald Pavia, G. Lampman et al. (Cengage Learning, 3rd edition or latest) (See last page of Syllabus for list of experiments)
4. ~~Chemistry 206 Lecture and Lab Notes*~~, by Alto Benedicto.
5. ~~Research Lab Notebook* (9 x 11, Hardbound, numbered pages, non-spiral permanent bound, graphed or lined pages).~~
6. ~~Chemistry Laboratory Coat* (white, long sleeve, knee length)~~
7. ~~Seven (7) Scantron Sheets Form No. 882-E (submitted on the second day of class) and pencil~~
8. ~~safety goggles* (\$5; needed on second class day), non-programmable scientific calculator (\$15–\$25), close-toed shoes~~
9. free access to “Online Tutoring” (online tutoring with a live person) via Canvas

* Available at **IVC Chemistry/STEM Club**.

RECOMMENDED MATERIALS:

1. *Study Guide with Student Solutions Manual for McMurry’s Organic Chemistry*, 9th Edition. (Amazon Rent: \$18.34)
2. ~~*Odyssey Molecular Explorer (Student Edition)*, by Wavefunction, Inc. (Molecular Modeling software)~~
3. Molecular Models* (*HGS Maruzen Organic Chem Kit #1003A \$32+tax*)

Course Requirements and Instructional Methods

1. Attendance for the entire class period is mandatory for Chem 206 Lab Classes. A Lab roll call will be initiated by the instructor within the first 5 minutes of Lab class. ~~If you are sent out during class (e.g., failure to~~

- comply with safety rules such as wearing Safety Goggles, etc.), you will be marked absent for that Lab, and will garner zero points for the experiment.
2. There are **no make-up Exams or Lab Classes**. A score of **zero (0)** will be recorded unless the absence is attributed to representation of official college functions. It is the student's responsibility to show proof of such function **prior** to the date of the absence.
 3. During Exam, the only things allowed are: **pencil, nonprogrammable calculator, and I.D.** You will be supplied with a Scantron. You may use the Exam Questionnaire as scratch paper. The Exam Questionnaire, and Scantron are to be submitted at the end of the Exam. **Possession of electronic devices (phones, ipod, programmable calculator, etc.) during Exam is considered cheating** and will be dealt with according to IVC policy.
 4. Each student is REQUIRED to **buy the Chem 206 textbook** and to **sign up for online HW (OWLv2 and LabSkills) no later than the second week of class**. Personal laptop is highly encouraged for online HW during Lab Class.
 5. **Due dates for Online Chapter HWs are found in the Class Schedule of Topics (see last page).**
 6. **Due dates for Quizzes are found in the Class Schedule of Topics (see last page).**
 7. **Due dates for Pre-lab Reading HW are found in the Class Schedule of Topics (see last page).**
 8. **Due dates for Worksheets (uploaded to Canvas as PDF file) are found in the Class Schedule of Topics (see last page). There is also Canvas Discussion that is graded. (Due date found in last page of Syllabus)**
 9. Prior to start of Lab Class, students are to fill out the Lab Notebook with INK with the following **Prelab Information: Date, Descriptive Title, Chemical Equation, Side Reactions, Table of Physical Constants, Calculations, Illustration of Apparatus Setup, Outline of Procedure**. Submit the notebook **within the first two minutes of class** for full pre-lab credit, therefore, don't be late!!! At times, Prelab quiz on said experiment will be given prior to start of the experiment.
 10. Before leaving the Lab Class, make sure the **instructor has signed** your Lab Notebook. Cross out mistakes with a single strike through line. Use appropriate verb tense. Cross out large blank areas in the notebook. Sign and date your notebook. **Notebook (containing Graphs, Spectroscopic Data, % Yield, etc. as need be) with answer to Post-Lab Questions are to be submitted within the first two minutes of the next time Lab meeting.**
 11. Products obtained from Labs must be submitted in a vial with the following information: **Your Name, Name of Compound, melting point and other relevant data, purity, yield in grams**. Points will be subtracted for missing information.
 12. Lab clean-ups are done 15 minutes before the end of lab. A **wet towel** should be used to wipe the lab bench in order to gain full points. Make sure sink and work area is clean. Points will be deducted to the entire class if the common work areas (fume hood, analytical balances) are dirty.
 13. There are no bonus work available. Kindly seek assistance immediately to clarify any questions.
 14. Keep up with the chapter readings. Seek help immediately on unclear concepts.
 15. If this is an online class, then the **deadlines are fixed**. Exams may be taken ahead of time but never after. Two choices to take Exam ahead of time: Choice A (12 hours ahead) and Choice B (24 hours ahead). Inform the instructor at least 24 hours of the new time, e.g., if Choice B, then inform instructor 48 hours ahead.

Out of Class Assignments: The Department of Education policy states that one (1) credit hour is the amount of student work that reasonably approximates not less than one hour of class time and two (2) hours of out-of-class time per week over the span of a semester. WASC has adopted a similar requirement.

Course Grading Based on Course Objectives

Assessment Type	How many	Total Points
Lecture Exams	5 @ 60	300 pts
Lecture Final Exam	1 @ 150	150 pts
Lab Final Exam	1 @ 150	150 pts
Quizzes	12 @ 10 pts 1 @ 15; 1 @ 5	140 pts
Online Homework	10 @ 10 3 @ 5	115 pts
Worksheets	9 @ 15	135 pts
Pre-lab HW	6 @ 5	30 pts
Labster Simulations	12 @ 10	120 pts
Canvas Discussion	5 @ 4 pts	20 pts

OVERALL POINTS = 1160 pts

Grading Scale Percentage	Letter Grade
85.00% to 100 %	A
75.00% to 84.99%	B
60.00% to 74.99%	C
50.00% to 59.99%	D

Attendance

- A student who fails to attend the first meeting of a class or does not complete the first mandatory activity of an online class will be dropped by the instructor as of the first official meeting of that class. Should readmission be desired, the student's status will be the same as that of any other student who desires to add a class. It is the student's responsibility to drop or officially withdraw from the class. See General Catalog for details.
- Regular attendance in all classes is expected of all students. **A student whose continuous, unexcused absences exceed the number of hours the class is scheduled to meet per week may be dropped.** For online courses, students who fail to complete required activities for two consecutive weeks may be considered to have excessive absences and may be dropped.
- Absences attributed to the representation of the college at officially approved events (conferences, contests, and field trips) will be counted as 'excused' absences.
- **Absences during Lab Classes, or leaving during Lab Classes automatically result in a grade of zero (0) for the Lab Experiment.**

Classroom Etiquette

- Electronic Devices: Cell phones and electronic devices must be turned off and put away during class, unless otherwise directed by the instructor.
- Food and Drink are prohibited in all classrooms. Water bottles with lids/caps are the only exception. Additional restrictions will apply in labs. Please comply as directed by the instructor.

- **Disruptive Students:** Students who disrupt or interfere with a class may be sent out of the room and told to meet with the Campus Disciplinary Officer before returning to continue with coursework. Disciplinary procedures will be followed as outlined in the [General Catalog](#).
- **Children in the classroom:** Due to college rules and state laws, no one who is not enrolled in the class may attend, including children.

Online Netiquette

- What is netiquette? Netiquette is internet manners, online etiquette, and digital etiquette all rolled into one word. Basically, netiquette is a set of rules for behaving properly online.
- Students are to comply with the following rules of netiquette: (1) identify yourself, (2) include a subject line, (3) avoid sarcasm, (4) respect others' opinions and privacy, (5) acknowledge and return messages promptly, (6) copy with caution, (7) do not spam or junk mail, (8) be concise, (9) use appropriate language, (10) use appropriate emoticons (emotional icons) to help convey meaning, and (11) use appropriate intensifiers to help convey meaning [do not use ALL CAPS or multiple exclamation marks (!!!)].

Academic Honesty

Academic honesty in the advancement of knowledge requires that all students and instructors respect the integrity of one another's work and recognize the important of acknowledging and safeguarding intellectual property.

There are many different forms of academic dishonesty. The following kinds of honesty violations and their definitions are not meant to be exhaustive. Rather, they are intended to serve as examples of unacceptable academic conduct.

- **Plagiarism** is taking and presenting as one's own the writings or ideas of others, without citing the source. You should understand the concept of plagiarism and keep it in mind when taking exams and preparing written materials. If you do not understand how to "cite a source" correctly, you must ask for help.
- **Cheating** is defined as fraud, deceit, or dishonesty in an academic assignment, or using or attempting to use materials, or assisting others in using materials that are prohibited or inappropriate in the context of the academic assignment in question.

Anyone caught cheating or plagiarizing will receive a zero (0) on the exam or assignment, and the instructor may report the incident to the Campus Disciplinary Officer, who may place related documentation in a file. Repeated acts of cheating may result in an F in the course and/or disciplinary action. Please refer to the [General Catalog](#) for more information on academic dishonesty or other misconduct. Acts of cheating include, but are not limited to, the following: (a) plagiarism; (b) copying or attempting to copy from others during an examination or on an assignment; (c) communicating test information with another person during an examination; (d) allowing others to do an assignment or portion of an assignment; (e) using a commercial term paper service.

Additional Student Services

Imperial Valley College offers various services in support of student success. The following are some of the services available for students. Please speak to your instructor about additional services which may be available.

- **Canvas Support Site.** The Canvas Support Site provides a variety of support channels available to students 24 hours per day.
- **Learning Services.** There are several learning labs on campus to assist students through the use of computers and tutors. Please consult your [Campus Map](#) for the [Math Lab](#); [Reading, Writing & Language Labs](#); and the [Study Skills Center](#).

- [Library Services](#). There is more to our library than just books. You have access to tutors in the [Study Skills Center](#), study rooms for small groups, and online access to a wealth of resources.

Disabled Student Programs and Services (DSPS)

Any student with a documented disability who may need educational accommodations should notify the instructor or the [Disabled Student Programs and Services](#) (DSP&S) office as soon as possible. The DSP&S office is located in Building 2100, telephone 760-355-6313. Please contact them if you feel you need to be evaluated for educational accommodations.

Student Counseling and Health Services

Students have counseling and health services available, provided by the pre-paid Student Health Fee.

- [Student Health Center](#). A Student Health Nurse is available on campus. In addition, Pioneers Memorial Healthcare District provides basic health services for students, such as first aid and care for minor illnesses. Contact the IVC [Student Health Center](#) at 760-355-6128 in Room 1536 for more information.
- [Mental Health Counseling Services](#). Short-term individual, couples, family, and group therapy are provided to currently enrolled students. Contact the IVC [Mental Health Counseling Services](#) at 760-355-6196 in Room 2109 for more information.

Student Rights and Responsibilities

Students have the right to experience a positive learning environment and to due process of law. For more information regarding student rights and responsibilities, please refer to the IVC [General Catalog](#).

Information Literacy

Imperial Valley College is dedicated to helping students skillfully discover, evaluate, and use information from all sources. The IVC [Library Department](#) provides numerous [Information Literacy Tutorials](#) to assist students in this endeavor.

Anticipated Class Schedule/Calendar

WK	DATE	CHAPTER READINGS	due dates Quiz due Tu 11:55 pm Pre-lab/Labster due Th 11:55 pm HW due Sat 11:55 pm	Due dates Exams on Wed 7 pm – 9 pm Worksheet due Fri 11:55 pm
1	Feb 13 – Feb 18	Review: Chap 17 and 18; Alcohols, Thiols, Sulfides	<i>Pre-lab A;</i> <i>HW17+;</i>	Safety Quiz in Webstar; Laboratory Check-in
2	Feb 20 – Feb 25	Ch 15: Benzene and Aromaticity	<i>Quiz 17+; Canvas Discussion #1</i> <i>Pre-lab B;</i> <i>HW15</i>	Worksheet 1 (Review of Chap 8,9,11,14,17)
3	Feb 27 – Mar 4	Ch 16: Benzene Derivatives: Electrophilic Aromatic Substitution	<i>Quiz 15;</i> <i>Pre-lab C;</i> <i>HW16</i>	Worksheet 2 (Chap 15, 16)
4	Mar 6 – Mar 11	Ch 19: Aldehydes and Ketones: Nucleophilic Addition Reactions	<i>Quiz 16;</i> <i>Pre-lab D; Labster 1 (Melting Point)</i> <i>HW19</i>	Lecture Exam 1 (covers Ch 15,16,17,18) Wed 7:00 pm – 9:00 pm

WK	DATE	CHAPTER READINGS	due dates Quiz due Tu 11:55 pm Pre-lab/Labster due Th 11:55 pm HW due Sat 11:55 pm	Due dates Exams on Wed 7 pm – 9 pm Worksheet due Fri 11:55 pm
5	Mar 13 – Mar 18	Ch 20: Carboxylic Acids and Nitriles	Quiz 19; <i>Canvas Discussion #2</i> Pre-lab E; Labster 2 Recrystallization HW20	Worksheet 3 (Chap 19, 20)
6	Mar 20 – Mar 25	Ch 21: Carboxylic Acid Derivatives: Nucleophilic Acyl Substitution.	Quiz 20; Pre-lab F; Labster 3 Aspirin Synthesis HW21	Worksheet 4 (Chap 21);
7	Mar 27 – Apr 1	Ch 22: Carbonyl Alpha-Substitution Reactions	Quiz 21; Labster 4 (Fractional Distillation) HW22	Lecture Exam 2 (covers Ch 18,19,20,21) Wed 7:00 pm – 9:00 pm
8	Apr 3 – Apr 8	Ch 23: Carbonyl Condensation Reactions	Quiz 22; <i>Canvas Discussion #3</i> Labster 5 (Liquid-liquid Extraction) HW 23	Worksheet 9 (Typewritten Paper for Lab 4)
9	Apr 10 -15	<i>Spring Break</i>	<i>Use this time to move ahead</i>	<i>Spring Break</i>
10	Apr 17 – Apr 22	Con't of Ch 23 and Ch 12: Mass Spectrometry	Quiz 23; Labster 6 (Thin-Layer Chrom) HW12	Worksheet 5 (Chap 23)
11	Apr 24 – Apr 29	Ch 24: Amines and Heterocycles	Quiz 12+; Labster 7 (Infrared Spectroscopy) HW24	Lecture Exam 3 (covers Ch 22, 23, 24) Wed 7:00 pm – 9:00 pm
12	May 1 – May 6	Ch 25: Carbohydrates	Quiz 24; <i>Canvas Discussion #4</i> Labster 8 (Proton NMR) HW25	Worksheet 6 (Chap 24, 25)
13	May 8 – May 13	Ch 26: Amino Acids, Peptides, and Proteins; Ch 27: Lipids	Quiz 25; Labster 9 (Carbon NMR) HW26+	Worksheet 7 (Lab aspect of Chap 12)
14	May 15 – May 20	Ch 28: Nucleic Acids Ch 29: Metabolic Pathways	Quiz 26+; Labster 10 (Mass Spectrometry) HW28+	Lecture Exam 4 (Ch 25, 26, 27) Wed 7:00 pm – 9:00 pm Worksheet 8 (Lab aspect of Chap 12, 13, 14)
15	May 22 – May 27	Ch 30: Pericyclic Reactions	Quiz 28; Labster 11: (Size Exclusion Chrom)	Lab Final Exam (150 pts) on May 24 Wed 7 pm – 9 pm
16	May 29 – Jun 3	Ch 31: Synthetic Polymers	Quiz 30+; <i>Canvas Discussion #5</i> Labster 12 (Identification of Org Cpd by Spectroscopy)	Lecture Exam 5 (covers Ch 28, 29, 30, 31) Wed 7:00 pm – 9:00 pm
17	Jun 5 – Jun 8		LEC FINAL EXAM (150 pt) on Jun 7 Wed at 7 am -9:15 pm (2 hr 15 min)	

Quizzes (designated as Quiz) are due every Tuesday at 11:55 pm

Lecture Exams are Wednesday from 7:00 pm – 9:00 pm

Canvas Discussion are due on Wed at 11:55 pm

Pre-lab HW (designated as Pre-lab) and Labster are due every Thursday at 11:55 pm

Worksheets are to be submitted Friday by 11:55 pm by uploading to Canvas as PDF file

Online Chapter HW (designated as HW) are due every Saturday at 11:55 pm