

Basic Course Information

Semester	Fall 2015	Instructor Name	Dr. Alejandro Cozzani
Course Title & #	Physics 202	Email	alex.cozzani@imperial.edu
CRN #	10065	Webpage (optional)	Refer to Blackboard
Room	2731	Office	2767
Class Dates	August 17 to December 11, 2015 Drop date: November 7, 2015.	Office Hours	Mondays through Thursday 7:00- 7:30 AM, M-W: 1:00-2:00 PM.
Class Days	Mondays and Wednesdays	Office Phone #	760-355-5720
Class Times	7:30-9:35 and 9:45 to 11:10 PM	Office contact if student will be out or emergency	Silvia Murray 760-355-6201 or Ofelia Duarte 760-355-6155
Units	5.0		

Course Description

This course is designed to give an understanding of the fundamental principles of physics in the areas of electricity, magnetism, atomic, and nuclear physics.

Prerequisite: PHYS 200 with a grade of "C" or better and MATH 194 with a grade of "C" or better or concurrent enrollment in MATH 194.

Student Learning Outcomes

1. Solve problems involving Gauss' Law.
2. Solve problems involving capacitors.
3. Solve problems involving magnetic fields in and near conductors, and the motion of charged particles in a magnetic field.
4. Solve problems involving EMF, resistor combinations, Kirchoff's Law.
5. Solve problems involving Faraday's and Lenz's Laws, and induced EMFs.

Course Objectives

1. The student will solve problems involving electric charges, electric field lines and the motion of a charged particle in a uniform electric field.
2. The student will solve problems involving Gauss' Law.
3. The student will solve problems involving electrical potential, potential energy due to point charges and continuous charge distributions.
4. The student will solve problems involving capacitors.
5. The student will solve problems involving current, resistance, electrical energy and power.
6. The student will solve problems involving EMF, resistor combinations, Kirchoff's Law, and RC circuits.
7. The student will solve problems involving magnetic fields in and near conductors, and the motion of charged particles in a magnetic field.
8. The student will solve problems involving the magnetic field of various sources.
9. Student will solve problems involving Faraday's and Lenz's Laws, and induced EMFs.
10. The student will solve problems involving inductance for RL, LC, and RLC circuits.
11. The student will solve problems involving resistors, inductors, and capacitors in an AC circuit.
12. The student will solve problems involving electromagnetic waves.
13. The student will solve problems involving molecular bonds, the energy spectra of molecules, and semiconductors.
14. The student will solve problems involving nuclear binding energy, radioactivity, and the decay process.

- The student will solve problems involving collisions between nuclear particles, fission, fusion, and elementary particles.

Textbooks & Other Resources or Links

- Textbook:** Fundamental of Physics, 10th edition, Chapters 21-32, 41-43, ISBN: 978-1-118-23072-5.
- Author:** Halliday & Resnick, by Jearl Walker.

Course Requirements and Instructional Methods

- Homework:** The purpose of homework is to provide the student with sufficient practice to master all topics studied in class and to do well on tests. Homework is done online at www.masteringphysics.com.
Course ID: [MPCOZZANI80797](#).
- Lab Reports:** These reports must be typed, double-space, font Times New Roman or similar, size 12, and the graphs must be done with Excel or any graphing program (i.e. TI InterActive). Reports are due a week after the specific experiment has been performed (If the experiment was done on September 03, it is due on September 10). No corrections will be allowed. No late submissions!
- Lecture Notes:** On lecture days, students are expected to have read the chapter in advance and bring some written notes to class (typed or handwritten) for discussion. No credit will be given but it is highly recommended.
- Tests or Exams:** They may be T/F, multiple choice or combination of T/F and/or multiple choice and free response questions. No makeup exams!
- Lab Tests:** Students will be tested on laboratory experiments. These will be based on the data collected and the analysis questions on the experiments. You may be asked the exact same questions or similar to those found on the lab manual and some theoretical questions related to those labs. No makeup exams!
- The laboratory environment contains a variety of chemical and physical hazards. It is vital to understand those potential hazards and their safeguards in order to prevent accidents and injuries. In order to work in a laboratory in the Department of Physics at Imperial Valley College, the student must understand and agree to abide by the laboratory safety rules set forth. Please log into Webstar with your credentials and find [Sports Survey and Safety Policy](#). Read the guidelines and answer yes to all the questions and click submit. Failure to comply will result in labs no participation with the corresponding zeros in experiments until the form is submitted.
- Mid-term:** It may include questions from the tests (recycled questions) and new questions (you have not seen them before but with similar difficulty). No makeup!
- Final Exam:** It may include questions from the tests (recycled questions) and new questions (you have not seen them before but with similar difficulty). The MC section will include ALL chapters. No makeup!
- Special Project:** Please see below.

Rubric

Criterion	High (5)	Medium (3)	Medium-Low (2)	Low (1)	Student Evaluation	Instructor Evaluation
Content/ information	accurate and concise; all relevant information is presented completely; clearly describes all principles involved; gives accurate history of application or theory	information is accurate; relevant information is present with some details missing; states all principles involved & describes most; gives brief history	information has some errors; most of the relevant information is present; states some of the principles covered; no history	major errors in information presented; not all relevant information presented; names a few or none of the principles involved; no history		
Presentation	makes eye contact;	some eye contact;	no eye contact;	avoids looking at		

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	<i>speaks knowledgeably without referring to notes; involves fellow students; clear well modulated voice</i>	<i>little need to reference notes; some involvement with fellow students; varies voice at times</i>	<i>uses notes frequently; very little involvement with fellow students; rarely varies voice</i>	<i>audience; reads notes; no involvement with fellow students; speaks in a monotone</i>		
Visual Aids (models, diagrams, etc.)	<i>aid used in the presentation is neat and organized; provides excellent support to the presentation making the words more easily understood</i>	<i>aid is used but as such is messy (globs of glue, dirty/cramped, dirty, pieces of tapes, etc.); provides good support for the presentation</i>	<i>visual aid is messy and poorly organized; adds little support to the presentation</i>	<i>no visual aids used</i>		
Creativity	<i>keeps other students interested throughout</i>	<i>some students appear distracted at times during the presentation</i>	<i>fails to capture and maintain interest of all students</i>	<i>fails to capture student interest at any time</i>		
Organization	<i>presentation follows a logical pattern; smooth transitions between sections</i>	<i>presentation follows a logical pattern; only a few rough points</i>	<i>presentation not given in a logical sequence but some organization present; transitions are abrupt</i>	<i>presentation lacks organization; speaker appears to move randomly from one idea to the next</i>		
Understanding of the Topic	<i>presenter conveys an outstanding understanding of the material</i>	<i>presenter conveys a good understanding of the material</i>	<i>presenter lacks a complete understanding of the material</i>	<i>presenter has a poor understanding of the material</i>		

Oral presentation: maximum 30 points

- a. Follow Rubric for point distribution.

Topics: Any chapters not addressed in class.

Review questions: 10 points

- b. Between 5 and 10.
- c. They should reflect what you have taught to your classmates. You may use the ones available in BB but make sure you know the answers and the reason for those answers.

Review problems: 10 points (about five with increasing level of difficulty).

- d. You have to be able to explain them to your classmates so they will understand.

Presentation dates: according to sign-up list. Once dates have been established, you cannot change it because presentations have a sequential order. You may pick the topic and your team members (no more than 3 per group) or you may work individually if you prefer to do so.

Minimum time is 30 minutes and up to an hour long.

- If you are absent the day of your presentation, your grade is ZERO (no exceptions!) so plan ahead.

- Use your own computer.

Course Grading Based on Course Objectives

The student's grade will depend on the following areas (not on total points):

➤ Homework	20%
➤ Tests – Presentation	20%
➤ Lab Reports - Lab Tests	20%
➤ Mid-term	20%
➤ Final Exam	20%
➤ TOTAL	100%

All grades are calculated by using the standard scale of:

A = 100-90% **B = 89-80%** **C = 79-70%** D = 69-60% F = 59% and below

Blackboard displays two grades: the weighted and the total. Your grade is the weighted one, so please keep it in mind.

Attendance

- A student who fails to attend the first meeting of a class or does not complete the first mandatory activity of an online class will be dropped by the instructor as of the first official meeting of that class. Should readmission be desired, the student's status will be the same as that of any other student who desires to add a class. It is the student's responsibility to drop or officially withdraw from the class. See General Catalog for details.
- Regular attendance in all classes is expected of all students. A student whose continuous, unexcused absences exceed the number of hours the class is scheduled to meet per week may be dropped. For online courses, students who fail to complete required activities for two consecutive weeks may be considered to have excessive absences and may be dropped.
- Absences attributed to the representation of the college at officially approved events (conferences, contests, and field trips) will be counted as 'excused' absences.

Classroom Etiquette

- Electronic Devices: Cell phones and electronic devices must be turned off and put away during class unless otherwise directed by the instructor.
- Calculators: scientific or graphing calculators can be used during class time and exams. NO phones or tablets as a substitute for calculators during exams.
- Food and Drink: are prohibited in all classrooms. Water bottles with lids/caps are the only exception. Additional restrictions will apply in labs. Please comply as directed.
- Disruptive Students: Students who disrupt or interfere with a class may be sent out of the room and told to meet with the Campus Disciplinary Officer before returning to continue with coursework. Disciplinary procedures will be followed as outlined in the General Catalog.
- Children in the classroom: Due to college rules and state laws, no one who is not enrolled in the class may attend, including children.

Academic Honesty

- Plagiarism is to take and present as one's own the writings or ideas of others, without citing the source. You should understand the concept of plagiarism and keep it in mind when taking exams and preparing written materials. If you do not understand how to correctly 'cite a source', you must ask for help.
- Cheating is defined as fraud, deceit, or dishonesty in an academic assignment or using or attempting to use materials, or assisting others in using materials, or assisting others in using materials, which are prohibited or inappropriate in the context of the academic assignment in question.

Anyone caught cheating or will receive a zero (0) on the exam or assignment, and the instructor may report the incident to the Campus Disciplinary Officer, who may place related documentation in a file. Repeated acts of cheating may result in an F in the course and/or disciplinary action. Please refer to the General School Catalog for more information on academic dishonesty or other misconduct. Acts of cheating include, but are not limited to the following: (a) plagiarism; (b) copying or attempting to copy from others during an examination or on an assignment ;(c) communicating test information with another person during an examination; (d) allowing others to do an assignment or portion of an assignment, (e) use of a commercial term paper service.

Additional Help

- Blackboard support center: <http://bbcrm.edusupportcenter.com/ics/support/default.asp?deptID=8543>
- Learning Labs: There are several 'labs' on campus to assist you through the use of computers, tutors, or a combination. Please consult your college map for the Math Lab, Reading & Writing Lab, and Learning Services (library). Please speak to the instructor about labs unique to your specific program
- Library Services: There is more to our library than just books. You have access to tutors in the learning center, study rooms for small groups, and online access to a wealth of resources.

Disabled Student Programs and Services (DSPS)

Any student with a documented disability who may need educational accommodations should notify the instructor or the Disabled Student Programs and Services (DSP&S) office as soon as possible. The DSP&S office is located in Building 2100, telephone 760-355-6313 if you feel you need to be evaluated for educational accommodations.

Student Counseling and Health Services

Students have counseling and health services available, provided by the pre-paid Student Health Fee. We now also have a fulltime mental health counselor. For information see <http://www.imperial.edu/students/student-health-center/>. The IVC Student Health Center is located in the Health Science building in Room 2109, telephone 760-355-6310.

Student Rights and Responsibilities

Students have the right to experience a positive learning environment and due process. For further information regarding student rights and responsibilities please refer to the IVC General Catalog available online at http://www.imperial.edu/index.php?option=com_docman&task=doc_download&gid=4516&Itemid=762

Information Literacy

Imperial Valley College is dedicated to help students skillfully discover, evaluate, and use information from all sources. Students can access tutorials at <http://www.imperial.edu/courses-and-programs/divisions/arts-and-letters/library-department/info-lit-tutorials/>

Anticipated Class Schedule / Calendar

WEEK # START DAY	CORE CONTENT	READING DUE	ASSIGNMENT DUE
1-August 17	Day 1: Syllabus / Introduction Chapter 21-part 1 Day 2: Chapter 21-part 2	Chapter 21	
2-August 24	Day 1: Chapter 22-part 1 Day 2: Chapter 22-part 2	Chapter 22	
3-August 31	Day 1: Chapter 23-part 1 Day 2: Chapter 23-part 2	Chapter 23	
4- September 07	Day 1: No class Day 2: Chapter 24-part 1	Chapter 24	
5- September 14	Day 1: Chapter 24-part 2 Day 2: Test # 1		Test # 1 (Chapters 21-22-23)
6- September 21	Day 1: Chapter 25-part 1 Day 2: Chapter 25-part 2	Chapter 25	
7- September 28	Day 1: Chapter 26-part 1 Day 2: Chapter 26-part 2	Chapter 26	
8-October 05	Day 1: Chapter 27-part 1 Day 2: Chapter 27-part 2	Chapter 27	
9- October 12	Review for Mid-term Mid-term (Chapters 21-26)	Chapters 21-26	
10- October 19	Day 1: Chapter 28-part 1 Day 2: Chapter 28-part 2	Chapter 28	
11- October 26	Day 1: Chapter 29-part 1 Day 2: Chapter 29-part 2	Chapter 29	

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12-November 02	Day 1: Chapter 30-part 1 Day 2: Test # 2	Chapter 30	Test # 2 (Chapters 27-28-29)
13- November 09	Day 1: Chapter 30-part 2 Day 2: Chapter 31	Chapter 31	
14- November 16	Day 1: Chapter 32 Day 2: Chapter 41	Chapter 32	Presentations
November 23	THANKSGIVING BREAK	NO CLASS	
15-November 30	Day 1: Chapter 42 Day 2: Chapter 43		Presentations
16-December 07	Day 1: Final Exam Day 2: Final Grades		Final Exam (Chapters 21-32)