



Basic Course Information

Semester:	Winter 2026	Instructor Name:	Dr. James Fisher
Course Title & #:	Chemistry 100 General Chemistry	Email:	Jim.fisher@imperial.edu
CRN #:	15018	Webpage (optional):	NA
Classroom:	2715	Office #:	2771
Class Dates:	Jan 5 – Feb 4 2025	Office Hours:	NA
Class Days:	Monday-Friday	Office Phone #:	760-355-6524
Class Times:	7:30 AM-12:15 PM	Emergency Contact:	Department Secretary 760-355-6155
Units:	4	Class Format/Modality:	Face to face

Course Description

Elementary principles of general inorganic chemistry, a preparatory course for CHEM 200. Previous science background is recommended but not required. This course is designed for non-science majors and students who need only a one-semester general chemistry course, and also for students entering a paramedical and allied health field, and industrial applications such as power plants. This course will satisfy the prerequisite for CHEM 200. (CSU) (UC credit limited. See a counselor.)

Course Prerequisite(s) and/or Corequisite(s)

Appropriate placement as defined by AB705; or
MATH 098 - Foundations of Algebra (Active); or
MATH 091 - Intermediate Algebra (Active) with a grade of "C" or better.

Student Learning Outcomes

Upon course completion, the successful student will have acquired new skills, knowledge, and or attitudes as demonstrated by being able to: Solve chemical problems using modern atomic theory. (ILO2, ILO4). Perform chemical experiments in a scientific manner using proper techniques, data analysis, and safety equipment. (ILO2, ILO3, ILO4).

Course Objectives

Upon satisfactory completion of the course, students will be able to:

1. calculate English and metric unit conversions and measurements using dimensional analysis.
2. write symbols for elements and know common ionic charges.
3. derive and write formulas and names for chemical compounds.
4. write and balance common chemical equations and identify reaction types.
5. solve stoichiometric problems, including their solutions using dimensional analysis.
6. describe atomic structure including isotopes, periodicity and molecular structure in terms of subatomic particles.

7. identify types of energy and calculate specific heat; identify energy involved in change of state including heat of vaporization and predict behaviors in cooling curves; calculate caloric and nutritional values of various foods.
8. describe gas behavior and solve problems involving the various gas laws.
9. identify the type of intermolecular forces existing between molecules, and its effect on macroscopic property of the substance.
10. calculate solution concentration of various types including dilutions.
11. define the three basic concepts (Arrhenius, Bronsted-Lowry and Lewis) of acids and bases and perform titration experiments and calculate pH.
12. use Le Chatelier's Principle to predict the shift in the direction of the reactants/products
13. determine the oxidant/reductant and balance redox equations.
14. describe nuclear processes and write nuclear equations using the subatomic particles involved and identify health factors and risks involved.

Textbooks & Other Resources or Links

1. Our online textbook [LibreTexts](#) Chemistry: Beginning Chemistry (Ball)
 - (Preferred) Textbook at LibreTexts is: Beginning Chemistry (Ball) [Introductory Chemistry - Chemistry LibreTexts](#)
 - At Canvas there are lectures PPT and videos
 - Exercises: are found in the above link.
2. [Lab Manual Chemistry 100 Laboratory](#) Manual; purchase from the STEM/Chem club, \$20.00.
3. 2@spiral ringed notebooks, 1 for notes taken while reading your textbook and 1 for working problems/exercises at the end of the chapter. These will be your study guides for the final.
4. Safety Glasses or Goggles: must be acid and heat resistant. These must comply with:
 - a. Meet ANSI* Z87.1-2003 standards.
 - b. Polycarbonate lens
 - c. Wraparound protection offers a wide field of vision.
5. Nonprogrammable Calculator: a highly recommended calculator is the Texas Instruments TI36X Solar Scientific Calculator (not the "TI36X-Pro" or the TI-30Xa).
6. Scranton for your final exam an 882-E, for 100 answers.
7. If you want to Zoom email me 1st and use: <https://imperial-edu.zoom.us/j/2198947419>

Course Requirements and Instructional Methods

- **Lecture Quizzes:** A short quiz on lecture material will periodically be given at the beginning of class. Quizzes are worth 5-15 points each with **no makeup** quizzes allowed. Quizzes will not be given on lecture exam days.
- **Lecture Exams:** Under normal circumstances (**Fall, Spring or 16-week semester**), there will be 5 exams and during Winter (4-week) or Summer (6-week) there will be 4 exams. There are no **make-up** exams unless you have a medical doctor notice. Exams will be graded and then returned as soon as possible. Only **non-programable** calculators are allowed.



- **Safety** in the laboratory is of utmost importance - those who do not follow the outlined safety procedures will either have points deducted from their lab score or asked to leave the lab during that lab. Closed toed shoes and goggles are required.
- **Laboratory:** To begin an experiment, the instructor must initial the pre-lab. This is necessary to ensure safety in the lab. In addition, each lab experiment will require data, calculations, and discussion writeup that is completed in your laboratory manual. There are no lab make-ups. Unless otherwise instructed, each student will work on experiments individually or as directed.
- **Completed experimental lab writeups** are due on due dates which are posted in your lab schedule.
- **Lab Exam:** A Lab exam will contain problems and/or explanation type questions based on the preceding laboratory experiments. Your Lab Notebook can be used during the Lab Exam. There is 1 Lab exams Fall and Spring, each of which count toward your course grade: no lab exams in Winter or Summer. No make-up Lab exams will be allowed. This Point Total is added to your Lecture Score to obtain a total score that includes both the lecture and lab component of this class. Only non-programable calculators are allowed.
- **Lab Cleanup** The entire class will lose points if the sinks, scales, hoods, floor are not clean, chemical caps not screwed back on, and chairs not put in place. The class can lose up to 10 points per lab.
- **Final Exam:** The Final Exam is comprehensive. Final exam questions are in multiple-choice format. You must purchase an 882-E, 50 questions per side, Scranton for the Final Exam. There are **no make-ups** because the date and time of the Final is the last day of class. Only non-programable calculators are allowed.
- **Out of Class Assignments:** The Department of Education policy states that one (1) credit hour is the amount of student work that reasonably approximates not less than one hour of class time and two (2) hours of out-of-class time per week over the span of a semester. WASC has adopted a similar requirement.
- **You must** (1) remember your locker combination-after locker check-in, (2) bring goggle or eye safety glasses, (3) closed toed shoes to be in the lab; you are not furnished these and (4) calculators for exams. Forgetting to do so will cost you 5 points.
- **Zoom face-to-face:** Zoom is for a video face-to-face instruction; we can work on homework problems etc. In Canvas, on the left banner is a link to "ConferZoom." Since this is new, and I don't know when students want to video/conference call, first contact me (Cranium Café/email) then we can initiate a Zoom Conference.
- **Fisher Zoom:** <https://cccconfer.zoom.us/j/9867876993>
- **Pronto** is on Canvas.

Course Grading Based on Course Objectives

- **Study Hints:** Chemistry is a very demanding course. Depending on your background, you will need to spend 1-4 hours outside of lab to get your work done. Missing a lecture usually means your grade falls by ½ grade.

- **Do not fall behind so:**
 - **Go to office hours**
 - **Get a tutor**
 - **Form study groups**
- **No Gifts, cards, or food; all will be refused. Spend your time and effort studying.**
- **Keep up, study everyday to avoid stress during finals week.**
- **Read and do problems every day, this makes it easier to memorize information.**

In-class quizzes	3 @ 5pts	15 pts
Exams	4 @ 100pts	400 pts
Labs	9 @ 10pts	90 pts
Lab Cleanup	9 @ -10pts ("-" if necessary)	
Locker Checkout		20 pts
Final Exam		200 pts
TOTAL (about)		+700 pts
- Letter grades will be assigned based upon the % of points earned: Grading scale, A: 90-100%; B: 80-89%, C: 70-79%, D: 60-69, F: <59.
- Grades are on Canvas

Academic Honesty (Artificial Intelligence -AI)

AI is allowed and encouraged in this class. This can be a tool to help you look up material while studying, or you can ask AI to generate questions/answers for you. Give AI questions and ask for detailed solutions.

IVC values critical thinking and communication skills and considers academic integrity essential to learning. Using AI tools as a replacement for your own thinking, writing, or quantitative reasoning goes against both our mission and academic honesty policy and will be considered academic dishonesty, or plagiarism unless you have been instructed to do so by your instructor. In case of any uncertainty regarding the ethical use of AI tools, students are encouraged to reach out to their instructors for clarification.

The following kinds of honesty violations and their definitions are not meant to be exhaustive. Rather, they are intended to serve as examples of unacceptable academic conduct.

- Plagiarism is taking and presenting as one's own the writings or ideas of others, without citing the source. You should understand the concept of plagiarism and keep it in mind when taking exams and preparing written materials. If you do not understand how to "cite a source" correctly, you must ask for help.
- Cheating is defined as fraud, deceit, or dishonesty in an academic assignment, or using or attempting to use materials, or assisting others in using materials that are prohibited or inappropriate in the context of the academic assignment in question.

Anyone caught cheating or plagiarizing will receive a zero (0) on the exam or assignment, and the instructor may report the incident to the Campus Disciplinary Officer, who may place related documentation in a file. Repeated acts of cheating may result in an F in the course and/or disciplinary action. Please refer to the [General Catalog](#) for more information on academic dishonesty or other misconduct. Acts of cheating include, but are not limited to, the following: (a) plagiarism; (b) copying or

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attempting to copy from others during an examination or on an assignment; (c) communicating test information with another person during an examination; (d) allowing others to do an assignment or portion of an assignment; (e) using a commercial term paper service.

Course Policies

Attendance

- A student who fails to attend the first meeting of a class or does not complete the first mandatory activity of a class will be dropped by the instructor as of the first official meeting of that class. Should readmission be desired, the student's status will be the same as that of any other student who desires to add a class. It is the student's responsibility to drop or officially withdraw from the class. See [General Catalog](#) for details.
- **Regular attendance in all classes is expected of all students. A student whose continuous, unexcused absence exceed the number of hours the class is scheduled to meet per week may be dropped; Chemistry 100, four units, is six hours and all other Chemistry courses, five units, are nine hours.** For online courses, students who fail to complete required activities for two consecutive weeks may be considered to have excessive absences and may be dropped.
- **Lab Attendance** is recorded just as lecture attendance. **If you miss the safety or introduction of the lab, first 5-10 minutes, you will not be able to attend that lab, and there are no lab makeups. You will receive no points for a lab you miss. Two (2) unexcused absences and you will be dropped. You may be asked to have your lab signed by the Instructor, at the beginning and end of the lab to receive any credit. Since Closed Toed Shoes are mandatory for Lab, not having closed toed shoes will not count as an absence, and you will NOT receive credit for the lab. Locker checkout counts as 2 labs or 20 points.**
- Absences attributed to the representation of the college at officially approved events (conferences, contests, and field trips) will be counted as 'excused' absences.

Classroom Etiquette

- Electronic Devices: Cell phones and electronic devices should be put in silent mode and put away during class.
- Food and Drink are prohibited in all classrooms. Water bottles with lids/caps are the only exception. Additional restrictions will apply in labs. Please comply as directed by the instructor.
- **Disruptive Students**: Students who disrupt or interfere with a class may be sent out of the room and told to meet with the Campus Disciplinary Officer before returning to continue with coursework. Disciplinary procedures will be followed as outlined in the [General Catalog](#).
- Children in the classroom: Due to college rules and state laws, no one who is not enrolled in the class may attend, including children.

IVC Student Resources

Student Services



Imperial Valley College offers various services in support of student success. The following are some of the services available for students. Please speak to your instructor about additional services which may be available.

- **CANVAS LMS.** Canvas is Imperial Valley College's main Learning Management System. To log onto Canvas, use this link: [Canvas Student Login](#). The [Canvas Student Guides Site](#) provides a variety of support available to students 24 hours per day. Additionally, a 24/7 Canvas Support Hotline is available for students to use: 877-893-9853.
- **[Learning Services](#).** There are several learning labs on campus to assist students through the use of computers and tutors. Please consult your [Campus Map](#) for the [Math Lab](#); [Reading, Writing & Language Labs](#); and the [Study Skills Center](#).
- **[Library Services](#).** There is more to our library than just books. You have access to tutors in the [Study Skills Center](#), study rooms for small groups, and online access to a wealth of resources.

Disabled Students Programs and Services (DSPS)

Any student with a documented disability who may need educational accommodations should notify the instructor or the [Disabled Student Programs and Services](#) (DSP&S) office as soon as possible. The DSP&S office is located in Building 2100, telephone 760-355-6313. Please contact them if you feel you need to be evaluated for educational accommodations.

Student Counseling and Health Services

Students have counseling and health services available, provided by the pre-paid Student Health Fee.

- **[Student Health Center](#).** A Student Health Nurse is available on campus. In addition, Pioneers Memorial Healthcare District provide basic health services for students, such as first aid and care for minor illnesses. Contact the IVC [Student Health Center](#) at 760-355-6128 in Room 1536 for more information.
- **[Mental Health Counseling Services](#).** Short-term individual, couples, family and group counseling services are available for currently enrolled students. Services are provided in a confidential, supportive, and culturally sensitive environment. Please contact the IVC Mental Health Counseling Services at 760-355-6310 or in the building 1536 for appointments or more information.

Veteran's Center

The mission of the [IVC Military and Veteran Success Center](#) is to provide a holistic approach to serving military/veteran students on three key areas: 1) Academics, 2) Health and Wellness, and 3) Camaraderie; to serve as a central hub that connects military/veteran students, as well as their families, to campus and community resources. Their goal is to ensure a seamless transition from military to civilian life. The Center is located in Building 600 (Office 624), telephone 760-355-6141.

Student Equity Program

- The Student Equity Program strives to improve Imperial Valley College's success outcomes, particularly for students who have been historically underrepresented and underserved. The college identifies strategies to monitor and address equity issues, making efforts to mitigate any disproportionate impact on student success and achievement. Our institutional data provides insight surrounding student populations who historically, are not fully represented. Student



Equity addresses disparities and/or disproportionate impact in student success across disaggregated student equity groups including gender, ethnicity, disability status, financial need, Veterans, foster youth, homelessness, and formerly incarcerated students. The Student Equity Program provides direct supportive services to empower students experiencing insecurities related to food, housing, transportation, textbooks, and shower access. We recognize that students who struggle meeting their basic needs are also at an academic and economic disadvantage, creating barriers to academic success and wellness. We strive to remove barriers that affect IVC students' access to education, degree and certificate completion, successful completion of developmental math and English courses, and the ability to transfer to a university. Contact: 760.355.5736 or 760.355.5733 Building 100.

- The Student Equity Program also houses IVC's Homeless Liaison, who provides direct services, campus, and community referrals to students experiencing homelessness as defined by the McKinney-Vento Act. Contact: 760.355.5736 Building 100.

Student Right

Students have the right to experience a positive learning environment and to due process of law. For more information regarding student rights and responsibilities, please refer to the IVC [General Catalog](#).



Anticipated Class Schedule/Calendar

Subject to change without prior notice

Introduction to Chemistry (15018) Monday thru Friday Lecture 7:300-9:45AM & Lab 10:00-12:15 Winter 2026 Room 2715					
WK	DAY	DATE	LECTURE	LABORATORY	Due Date
1	Mon.	1-5	Chapters 1	Procedures, Safety, and Equipment Policy	
	Tues.	1-6	Chapters 1 & 2	Lab 01 Measurements	
	Wed.	1-6	Chapters 2 & 3	Lab 01 Measurements	1-9
	Thurs.	1-8	Chapters 3 & 4	02_Nomenclature	
	Fri.	1-9	Chapters 3 & 4	Exam 1	
2	Mon.	1-12	Chapter 4	Lab 02 Nomenclature	1-14
	Tues.	1-13	Chapters 4 & 5	Lab 03_Controlling Limiting Reactants (CRL)	1-16
	Wed.	1-14	Chapter 5	Lab 04 NIE and 03 CLR	
	Thurs.	1-15	Chapters 5 & 6	Lab 04 NIE	1-20
	Fri.	1-16	Chapters 6 & 7	Exam 2	
3	Mon.	1-19	Holiday: Martin Luther King day (MLK)		
	Tues.	1-20	Chapter 7 & 8	Lab 05_Atoms, Molecules and Ions	1-23
	Wed.	1-21	Chapters 8 & 9	Lab 05 and Lab 06_VSEPR	
	Thurs.	1-22	Chapter 9	Lab 06_VSEPR	1-26
	Fri.	1-23	Chapters 10	Exam 3	
4	Mon.	1-26	Chapter 10	Lab 07_Properties of Gases	1-30
	Tues.	1-27	Chapters 10 & 11	Lab 08_Equilibrium UV-Vis	2-2
	Wed.	1-28	Chapters 11 & 12	Lab 08 and Lab 09_Titration	
	Thurs.	1-29	Chapter 12	Lab 09_Titration	2-2
	Fri.	1-30	Chapters 13 & 14	Exam 4	
5	Mon.	2-2	Chapter 14	Grade Lab notebooks	
	Tues.	2-3	Chapters 15 & 16	Lab Exam	
	Wed.	2-4		Final Exam	